

Sintesis ligan para di 2 1 methyl 3 pyridyl 4 5 dihydro 1h pyrazol 5 yl benzena sebagai sensor ion logam pb 2 = Synthesis of para di 2 1 methyl 3 pyridyl 4 5 dihydro 1h pyrazol 5 yl benzene ligand as sensor for pb 2 metal ions / Tirta Angen Pangestu

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Deskripsi Lengkap: <https://lib.ui.ac.id/detail?id=20412719&lokasi=lokal>

Abstrak

[ligan para-di-%-+!-rnmethyl-(-pyridyl-*,3-dihydro-!:-pyra<ol-3-yl/ben<ena sudah berhasil disintesis rnelalui rnetode kondensasi Aldol. :asil yang diperoleh dikarakterisasi rnellgunakan spektro2otorometer .>->is, spektro2otorometer in2rarnerah dan spektrotrneter NM?. Aplikasi ligan ini adalah sebagai Sensor ion logam Pb%= dan dilakukan rnellgunakan spektrotrneter .>->is. :asil dari studi sensor rnenunjukkan bahwa ligan rnermpunyai intensitas absorbansi yang kuat. :al ini didukung oleh nilai absorpti\$itas rncolar +@/ yang besar. Studi spektroskopi .>->is pada penarnbahan ion Pb%= rnenunjukkan rnunculnya puncak baru pada daerah panjang gelombang %A7 nrm. :al ini rnenunjukkan adanya pernentukan kornpleks antara ion logam +Pb%=/ dengan ligan para-di-%-+!-rnmethyl-(-pyridyl-*,3-dihydro-!:-pyra<ol-3-yl/ben<ena. Studi aplikasi sensor rnenunjukkan bahwa ligan ini dapat dijadikan sensor untuk ion Pb%= karena penarnbahan ion ini rnenyebabkan penurunan intensitas serapan absorbansi. :asil studi e2ekti\$itas sensor rnenunjukkan bahwa ligan para-di-%-+!-rnmethyl-(-pyridyl-*,3-dihydro-!:-pyra<ol-3-yl/ben<ena rnerupakan sensor yang e2ekti2 terhadap penarnbahan ion Pb%= pada panjang gelombang rnaksirnurn +Brnaks/ %A!

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19; 2; 581; 580; 591; 589; 623;
 613; 641; 609; 588; 581; 601; The ligand para-di-*m*-*n*-methyl-(*pyridyl*-,3-dihydro-*pyrazol-3-yl*/benzene
 has been synthesized well by using Aldol condensation method. The result of
 synthesis is characterized by visible spectrophotometer, infrared
 spectrophotometer and NMR spectrometer. The application of this ligand as
 sensor for Pb²⁺ metal ions was studied by using visible spectrophotometer.
 Sensing studies indicate that the ligand has a strong absorbance intensity. This is
 supported by a large molar absorptivity value. UV-Vis spectroscopy studies on
 the addition of Pb²⁺ ions showed the emergence of a new peak at 470 nm
 wavelength region. This indicates that the complex formed between metal ions
 +Pb²⁺ with para-di-*m*-*n*-methyl-(*pyridyl*-,3-dihydro-*pyrazol-3-yl*/benzene
 ligand. Application sensor studies showed that these ligands can be used as sensor
 for Pb²⁺ ions due to the addition of these ions causes a decrease in absorbance
 intensity. The results of the study of sensor selectivity showed that the ligand para-di-*m*-*n*-
 methyl-(*pyridyl*-,3-dihydro-*pyrazol-3-yl*/benzene is selective for
 addition of Pb²⁺ ions at the maximum wavelength 470 nm.]